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CO-1-2019

FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B. Pharmacy (First Year) (First Semester) EXAMINATION

MARCH/APRIL, 2019

HUMAN ANATOMY AND PHYSIOLOGY-I

Paper (BPLO-IT)

(Monday, 22-4-2019) (BP-101-T) Time: 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

- N.B. :- (i) All questions are compulsory.
(ii) Figures to the right indicate full marks.
(iii) Draw neat labelled diagrams wherever necessary.

1. Answer all the questions : 10×2=20

- (a) Define anatomy and physiology.
- (b) Give functions of Endoplasmic reticulum.
- (c) Define tissue. Enlist its types.
- (d) Give classification of bones.
- (e) Enlist the contractile protein of skeletal muscle.
- (f) Give composition of blood.
- (g) Draw neat labelled diagram of Lymph node.
- (h) Give classification of peripheral nervous system.
- (i) Differentiate between arteries and veins.
- (j) Write normal value and life span of leukocytes.

2. Answer the following (any two) : 2×10=20

- (a) Define blood pressure. Discuss in detail short-term and long-term mechanism involved in regulation of blood pressure.

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- (b) What is coagulation ? Discuss in detail about intrinsic and extrinsic pathway of blood coagulation.
- (c) Draw neat labelled diagram of plasma membrane. Describe in detail about active and passive transport mechanism.
3. Answer the following (any seven) : 7×5=35
- (a) What is Homeostasis ? Explain negative feedback mechanism with example.
- (b) Explain in detail about anatomy and physiology of skin.
- (c) Discuss about physiology and muscle contraction.
- (d) Write about anatomy and physiology of epithelial tissue.
- (e) Explain about bones of appendicular skeleton.
- (f) Explain in detail synovial joint.
- (g) What is erythropoiesis ? Explain the steps in erythropoiesis.
- (h) Write anatomy and physiology of eye.
- (i) Write about conducting system of heart.

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CO—05—2019

FACULTY OF PHARMACEUTICAL SCIENCE

B. Pharmacy (First Year) (First Semester) EXAMINATION

MARCH/APRIL, 2019

PHARMACEUTICAL ANALYSIS-I

(Wednesday, 24-4-2019) (CBP-102-T) Time : 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Answer to the point only.

(iii) Figures to the right indicate full marks.

1. Answer the following :

20

- (a) Define molarity and normality.
- (b) Give applications of polarography.
- (c) How to prepare conductivity water ?
- (d) Sketch a neat labelled diagram of silver chloride electrode.
- (e) Define oxidising agent and reducing agent.
- (f) Give difference between co-precipitation and post-precipitation.
- (g) Enlist types of non-aqueous solvents.
- (h) Define the term accuracy and precision.
- (i) Enlist the name of indicators used in non-aqueous titration.
- (j) What is masking and demasking agent.

2. Solve any two of the following :

20

- (a) What are mixed indicators ? Explain theories of acid-base indicators.
- (b) Define precipitation titration. Explain Mohr's method.
- (c) Explain in short electrochemical cell. Give construction and working of calomel electrode.

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3. Solve any *seven* of the following :

- (a) Define primary and secondary standards. Give ideal requirements for primary standard substances.
- (b) Give the procedure for preparation and standardisation of sodium hydroxide solution.
- (c) Write estimation of sodium benzoate.
- (d) Define non-aqueous titration. Classify non-aqueous solvents with example.
- (e) Describe steps involved in gravimetric analysis.
- (f) Give applications of conductometric titration.
- (g) Give construction and working of dropping mercury electrode.
- (h) Discuss Ilkovic equation.
- (i) Write the classification of complexometric titration.

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CO—09—2019

FACULTY OF PHARMACEUTICAL SCIENCES AND TECHNOLOGY

B.Pharm. (First Year) (First Semester) EXAMINATION

MARCH/APRIL, 2019

PHARMACEUTICS—I

(Friday, 26-4-2019) (BP-103-T) Time : 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

- N.B. :-*
- (i) All questions are compulsory.
 - (ii) Figures to the right indicate full marks.
 - (iii) Answer to the point only.

1. Solve the following : 10×2=20
- (a) Define Drug and Dosage form.
 - (b) Give importance of Pharmacopoeia.
 - (c) Mention different systems of weights and measures.
 - (d) Why glycerine is used as a base in throat paint ?
 - (e) What is the duty of pharmacist in case the medicine is prescribed in over dose ?
 - (f) Write the equation of Stokes law.
 - (g) Define ointment. How does it differ from pastes ?
 - (h) Give Young's and Dilling's formula for calculation of dose in children.
 - (i) Define :
 - (i) Throat paints
 - (ii) Elixirs
 - (j) Give advantages of suppositories.

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(2)

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2×10=20

2. Solve any *two* of the following :

- (a) Define and classify in detail sterile and non-sterile dosage forms.
- (b) What is incompatibility ? Explain in detail chemical incompatibility.
- (c) Define emulsions. Discuss in brief methods of preparation and stability parameters for emulsions.

3. Solve any *seven* of the following :

7×2=35

- (a) Write in brief about superscription and inscription in prescription.
- (b) Describe in brief history of profession of pharmacy in India.
- (c) Define Posology. Write in brief about :
 - (i) Synergism
 - (ii) Idiosyncrasy
 - (iii) Tachyphylaxis.
- (d) Define suspensions. Give advantages of suspensions.
- (e) Give evaluation of semisolid dosage form.
- (f) Give formula and method of preparation of Mouthwash.
- (g) Define suppositories. Write in brief cold compression method.
- (h) Write a note on simple and compound powders.
- (i) Describe different excipients used in formulation of liquid dosage form.

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CO—13—2019

FACULTY OF SCIENCE AND TECHNOLOGY

B. Pharmacy (First Year) (First Semester) EXAMINATION

MARCH/APRIL, 2019

PHARMACEUTICAL INORGANIC CHEMISTRY

(Monday, 29-4-2019) (CBP-104T) Time: 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

- N.B. :-**
- (i) All questions are compulsory.
 - (ii) Figures to the right indicate full marks.
 - (iii) Answer to the point only.

1. Answer the following

20

- (a) Sketch a neat labelled diagram of limit test for arsenic.
- (b) Define buffer capacity.
- (c) Give composition of ORS.
- (d) Why calcium compounds as antacid are usually administered with magnesium salt.
- (e) Name two inorganic substances used as antimicrobial agents.
- (f) Give molecular formula and molecular weight of potash alum.
- (g) What are expectorants? How do they act?
- (h) Give properties of alpha rays.
- (i) Define emetics with example.
- (j) Give uses of I^{131} .

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(2)

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2. Answer any *two* of the following :

- (a) Discuss in detail about physiological acid base balance.
- (b) Give ideal properties of antacids. Write properties and uses of Aluminium hydroxide gel.
- (c) What is anaemia ? How would you treat it ? Give method of preparation, properties and uses of ferrous sulphate.

3. Solve any *seven* of the following

35

- (a) Give principle and procedure of limit test for sulphate.
- (b) Give properties and assay of calcium gluconate.
- (c) Define saline cathartics. Give properties and uses of magnesium sulphate.
- (d) Write mechanism of action of antimicrobials.
- (e) What are astringents ? Give method of preparation and uses of zinc sulphate.
- (f) Define Acidifying agents. Write properties and uses of Dil. Hydrochloric acid.
- (g) Write a role of fluoride in the treatment of dental caries.
- (h) Describe two methods of measurement of radioactivity.
- (i) Write pharmaceutical applications of radioactive substances.

CO-13-2019

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CO—03—2019

FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B.Pharmacy (First Year) (Second Semester) EXAMINATION

MARCH/APRIL, 2019

HUMAN ANATOMY AND PHYSIOLOGY-II

[BP-201T]

(Tuesday, 23-4-2019)

Time : 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Write to the point only.

(iii) Figures to the right indicate full marks.

1. Solve all the following :

10×2=20

- (a) What are neurotransmitters ?**
- (b) Define absorption and digestion.**
- (c) Give role of ATP.**
- (d) Enlist upper and lower respiratory organ.**
- (e) Draw a neat labelled diagram of nephron.**
- (f) Write role of parathyroid hormone.**
- (g) Define menarche and menopause.**
- (h) Give composition of urine.**
- (i) Define peptic ulcer ?**
- (j) Give composition and functions of pancreatic juice.**

2. Solve any two of the following :

2×10=20

- (a) Draw neat labelled diagram of brain. Explain anatomy and physiology of cerebral cortex.**
- (b) Describe in detail anatomy and physiology of thyroid hormone.**
- (c) Discuss in detail about female reproductive system.**

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(2)

CO-03-2019

7×5=35

3. Solve any *seven* of the following :

- (a) Discuss about oogenesis.
- (b) Discuss about importance of Genetics.
- (c) Write anatomy and physiology of adrenal gland.
- (d) Give anatomy and physiology of liver.
- (e) Write anatomy and physiology of Hypothalamus.
- (f) Draw neat labelled diagram of neuron. Discuss in detail the mechanism of conduction of nerve impulses across nerve fibre.
- (g) Discuss about various phases of gastric secretion.
- (h) Write about various phases of menstrual cycle.
- (i) Write about Renin-angiotensin system.

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CO—7—2019

FACULTY OF SCIENCE AND TECHNOLOGY
B.Pharmacy (Second Semester) EXAMINATION
MARCH/APRIL, 2019

PHARMACEUTICAL ORGANIC CHEMISTRY-I

Paper BP-202-T

(Thursday, 25-4-2019)

Time : 10.00 a.m. to 1.00 p.m.

Time—Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw structure(s) and write reaction(s) wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer all the questions :

10×2=20

(a) What is Saytzeff Rule ?

(b) Draw the structure of :

(i) 3-Methylbutanamide

(ii) 5-Bromo-4-methyl-hex-3-en-2-one.

(c) What is Walden inversion ?

(d) Draw the structure and give the uses of the following :

(i) Chlorobutanol

(ii) Amphetamine.

(e) Arrange the following compounds in order of increasing acidity :

(i) 2-Fluoropropanoic acid

(ii) 2-Chloropropanoic acid

(iii) Propanoic acid

(iv) 2-Bromopropanoic acid.

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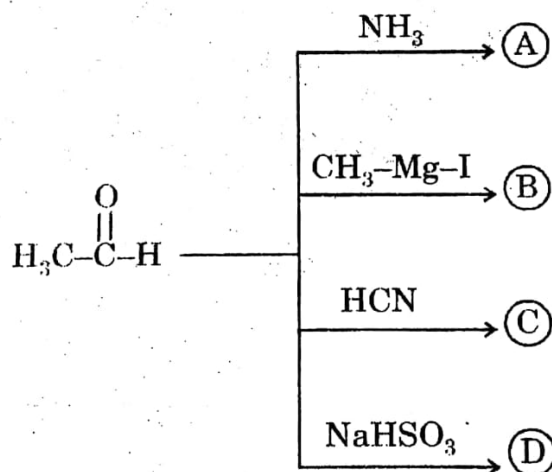
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- (f) How will you prepare 2-Butene from 1-Butene ? Write reaction.
- (g) Define isomerism. Draw the structure of various isomers of molecular formula C_3H_8O .
- (h) Explain why ketones are less reactive than aldehydes.
- (i) What is electromeric effect ?
- (j) Discuss the basicity of amines.

2. Solve any *two* of the following :

2×10=20

- (a) What is electrophilic addition to olefins ? Discuss addition reactions to olefins like hydrogenation, halogenation and ozonolysis. Discuss about peroxide effect.
- (b) What are alkyl halides ? How are they classified ? Explain in detail S_N1 and S_N2 reaction mechanism with stereochemistry.
- (c) (i) Complete the reaction :

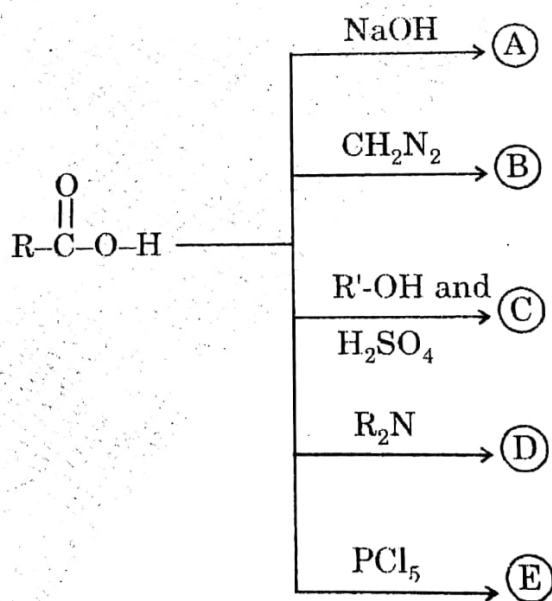


Write structure and name of A, B, C and D.

- (ii) Discuss about Perkin condensation and crossed Cannizzaro reaction.

3. Solve any seven of the following :

- (a) How will you prepare primary, secondary and tertiary alcohol and amines from Grignard's reagent ?
- (b) Define conjugated dienes. What is Diels-Alder reaction ? Discuss 1, 2 and 1, 4-addition reaction of 1, 3-Butadiene.
- (c) What is organic chemistry ? Give its importance. Classify organic compounds with a suitable example.
- (d) What is hybridization ? Explain hybridization of alkanes in detail. Comment on "Halogenation of alkanes".
- (e) Complete the following reaction and identify A, B, C, D and E with structure :



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(4)

- (f) Draw the structure and give the uses of :
- Ethanolamine
 - Vanilin
 - Acetyl salicylic acid
 - Dichloromethane
 - Methyl salicylate.
- (g) (i) Explain inductive effect in detail.
 (ii) Enlist the various test to distinguish between 1°, 2° and 3° alcohols. Explain any *two* tests.
- (h) Write any *four* chemical reactions of amines with mechanism of any *one*.
- (i) (1) Write the IUPAC name of :
- (a)
$$\begin{array}{c} \text{H}_2\text{C} = \text{CH} - \text{CH} - \text{COOH} \\ | \\ \text{CH}_3 \end{array}$$
- (b)
$$\begin{array}{c} \text{CH}_3 \\ | \\ \text{H}_3\text{C} - \text{C} - \text{CH}_2 - \text{CHO} \\ | \\ \text{C}_2\text{H}_5 \end{array}$$
- (2) How will you achieve the following synthetic conversions :
- Propene to 2-Methyl propanoic acid
 - Acetaldehyde to 1-Butanol
 - Ethanol to Acetylene.

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CO—11—2019

FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharmacy (First Year) (Second Semester) EXAMINATION

MARCH/APRIL, 2019

BIOCHEMISTRY

(BP203T)

(Saturday, 27-4-2019)

Time : 10.00 a.m. to 1.00 p.m.

Time— Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

(iii) Answer to the point only.

1. Solve *all* the questions :

20

(a) What are biomolecules ? Enlist any *two*.

(b) Differentiate between purines and pyrimidines.

(c) What are fatty acids ? Write its function.

(d) What is fatty liver ?

(e) Define free energy and redox potential.

(f) Write the structure and biochemical functions of cyclic AMP.

(g) What are essential and non-essential amino acids.

(h) Write the properties of enzymes.

(i) What are the functions of RNA ?

(j) Give the biological role of carbohydrates.

2. Answer any *two* of the following :

20

(a) Explain the structures of DNA and write biological importance of DNA.

(b) Explain the process of β -oxidation of fatty acids.

(c) Name the various pathways of glucose metabolism. Give in detail about TCA cycle.

P.T.O.

3. Solve any *seven* of the following :

(a) Write short notes on :

(i) Transamination

(ii) Deamination

(iii) Decarboxylation with example.

(b) Write a short note on energy rich compounds.

(c) Describe hormonal regulation of blood glucose level.

(d) Discuss about inhibitors of ETC [Electron Transport Chain] and oxidative phosphorylation.

(e) Enlist factors affecting on enzyme activity. Explain any *two*.

(f) Explain glycogen metabolism pathway.

(g) Write short notes on :

(i) Atherosclerosis

(ii) Hypercholesterolemia

(h) Explain process of protein synthesis.

(i) Write therapeutic and diagnostic applications of enzyme.

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CO—15—2019

FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharm. (First Year) (Second Semester) EXAMINATION

MARCH/APRIL, 2019

PATHOPHYSIOLOGY

(BP-204-T)

(Tuesday, 30-4-2019)

Time : 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw neat labelled diagram wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer the following questions :

10×2=20

- (a) Define adaptation and homeostasis.
- (b) What is calcification ?
- (c) What are Metaplasia and dysplasia.
- (d) Mention cardinal sign of inflammation.
- (e) Enlist the various inflammatory mediators.
- (f) What is Thalassemia.
- (g) What are hypo and hypothyroidism.
- (h) Write clinical manifestation of UTI.
- (i) What is Gout ?
- (j) Give causes of syphilis and Gonorrhoea.

2. Answer the following (any two) :

2×10=20

- (a) Describe in detail etiopathogenesis, clinical manifestation and treatment of AIDS.
- (b) Explain in detail etiopathogenesis, clinical manifestation and treatment of viral hepatitis and IBD.
- (c) Discuss in detail pathogenesis of cell injury.

P.T.O.

3. Answer the following (any seven) :

- (a) Explain basic mechanism involved in inflammation.
- (b) Write etiopathogenesis of diabetes mellitus.
- (c) Explain pathophysiology and treatment of Atherosclerosis.
- (d) What is Asthma ? Write its pathogenesis and clinical manifestation.
- (e) Write etiopathogenesis of epilepsy.
- (f) Write etiology, pathogenesis, clinical manifestation and treatment of Rheumatoid Arthritis.
- (g) Write etiology and pathogenesis of cancer.
- (h) Explain pathophysiology and treatment of Tuberculosis.
- (i) Explain pathogenesis and treatment of Parkinson's disease.