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CL-13-2019

FACULTY OF SCIENCE AND TECHNOLOGY

Pharm. D. (First Year) EXAMINATION APRIL/MAY, 2019

PHARMACEUTICAL INORGANIC CHEMISTRY

(Thursday, 2–5-2019)

Time: 10.00 a.m. to 1.00 p.m.

Time— Three Hours

Maximum Marks—70

N.B. := (i) All questions are compulsory.

- (ii) Answer to the point only.
- (iii) Figures to the right indicate full marks.
- 1. Solve any five of the following:

 $5 \times 2 = 10$

- (a) Write characteristics of random error.
- (b) Differentiate between iodometry and iodimetry.
- (c) Name two examples of fluorescent indicators.
- (d) Write disadvantages of gravimetric analysis.
- (e) Why is nitric acid used in limit test of chloride?
- (f) Give classification of acidifiers.
- 2. Solve any two of the following:

 $2 \times 6 = 12$

- (a) Define secondary standards. Give the requirements of primary standards with suitable example.
- (b) Explain in detail the following aqueous acid-base titrations with graphs.
 - (i) Strong acid with strong base,
 - (ii) Weak acid with strong base.
- (c) Classify solvents used in non-aqueous titration and give examples.

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3. Solve any *two* of the following:

 $2 \times 6 = 12$

- (a) Describe in detail about Fajan's method.
- (b) What are types of EDTA titration? Give procedure for standardization of EDTA.
- (c) Write principle and method for limit test of lead.
- 4. Solve any two of the following:

 $2 \times 6 = 12$

- (a) Write ideal properties of antacids.
- (b) Define cathartics. Write method of preparation and uses of magnesium sulphate.
- (c) Discuss biological role and uses of copper in human body.
- 5. Solve any *two* of the following:

 $2 \times 6 = 12$

- (a) Describe physiological acid-base balance.
- (b) Write method of preparation, physical properties and uses of potassium permanganate.
- (c) Discuss ideal properties of excipients used in medicaments.
- 6. Solve any *two* of the following:

 $2 \times 6 = 12$

- (a) What is dental caries? Write method of preparation and pharmaceutical uses of sodium fluoride.
- (b) Define haematinics. Give biological importance of iron.
- (c) Write applications of radioisotopos in medicines.