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CL-10-2019

FACULTY OF PHARMACEUTICAL SCIENCES AND TECHNOLOGY D Pharm. (First Year) EXAMINATION

MARCH/APRIL, 2019

PHARMACEUTICAL ORGANIC CHEMISTRY

(Monday, 29-4-2019)

Time: 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—70

- N.B. := (i) Figures to the right indicate full marks.
 - (ii) Draw chemical structures and write chemical reactions wherever necessary.
- 1. Solve any five of the following:

 $5 \times 2 = 10$

- (a) Give reason: Phenol is more acidic than alcohol.
- (b) What is Markonikov's rule?
- (c) Write medicinal uses of Vanillin.
- (d) Define solubility and B.P.
- (e) What is dehydration of alcohol? Write any one example.
- (f) What is mechanism of benzoin condensation?
- (g) Define oxidizing and reducing agents with suitable example.
- 2. Answer any two of the following:

 $2 \times 6 = 12$

- (a) Write definition, mechanism and applications of Hofmann rearrangement reaction.
- (b) Explain in detail Lewis theory and Lowry-Bronsted theory of acid and bases.
- (c) Write method of preparation, assay procedure and uses of aspirin.

P.T.O.

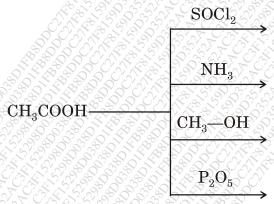
3. Solve any *two* of the following:

 $2 \times 6 = 12$

- (a) What are free radicals? Explain the mechanism of free radical chain reactions.
- (b) Explain mechanism and stereochemistry of S_N1 reactions.
- (c) Explain directive, activating and deactivating influence of substituents on benzene ring.
- 4. Answer any *two* of the following:

 $2 \times 6 = 12$

- (a) Explain in detail stability of carbocations.
- (b) Complete the following reactions:



- (c) Write mechanism of nitration and sulphonation reaction.
- 5. Answer any two of the following:

 $2 \times 6 = 12$

- (a) Explain Bayer-Strain theory in detail.
- (b) Write mechanism of crossed aldol condensation and Cannizzaro reaction.
- (c) Match the following pairs:

	Types of isomers		Examples
(i)	Nuclear	(a)	methyl propyl ether and diethyl ether
(ii)	Position	(<i>b</i>)	crotonic acid and isocrotonic acid
(iii)	Metamerism	(c)	diethyl ether and butyl alcohol
(iv)	Functional	(d)	(+) & (-) tartaric acid
(v)	Geometrical	(e)	<i>n</i> -butane and isobutane
(vi)	Optical	<i>(f)</i>	o-xylene and p-xylene

6. Answer any two of the following:

 $2 \times 6 = 12$

- (a) What is basicity of amines? Explain effect of substituents on basicity of amines.
- (b) Write addition reactions and free radical reactions of dienes.
- (c) Write IUPAC names of the following organic compounds:

$$(i)$$
 $CH_2 = CH-CH_2-CH_2-CH_3$

$$\stackrel{\cdot}{(iii)}\quad \stackrel{\cdot}{\mathrm{CH}_2}\!\!-\!\!\!\mathrm{CH}_2\!\!-\!\!\!\mathrm{COOH}$$

$$(iv) \qquad \begin{matrix} \text{OH} \\ \\ \text{O}_2\text{N} \\ \\ \text{NO}_2 \end{matrix}$$

$$(v) \qquad \text{CH}_3\text{--C}\text{--O.CH}_2\text{CH}_3$$

$$(vi) \quad \mathrm{CH_3-\!\!-}\mathrm{CH_2-\!\!-}\mathrm{NH-\!\!-}\mathrm{CO-\!\!-}\mathrm{CH_3}$$